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In organic chemistry, amines (/ ˈɛm ɪn, ˈɛn /, UK also / ˈeɪ m ɪn /) are compounds and functional groups that contain a basic nitrogen atom with a lone pair. Amines are formally derivatives of ammonia, wherein one or more hydrogen atoms have been replaced by a substituent such as an alkyl or aryl group (these may respectively be called alkylamines and arylamines; amines ...

Amine - Wikipedia

This text is an unbound, three hole punched version. Designed for courses in Cell Biology offered at the Sophomore/Junior level, Karp's Cell and Molecular Biology: Concepts and Experiments, Binder Ready Version, 8th Edition continues to be the best book in the market at connecting key concepts to the experiments that reveal how we know what we know in the world of Cell Biology.

Cell and Molecular Biology, Binder Ready Version: Concepts

Infrared spectroscopy exploits the fact that molecules absorb frequencies that are characteristic of their structure. These absorptions occur at resonant frequencies, i.e. the frequency of the absorbed radiation matches the vibrational frequency. The energies are affected by the shape of the molecular potential energy surfaces, the masses of the atoms, and the associated vibronic coupling.

Infrared spectroscopy - Wikipedia

Dieckmann condensation
 Dieckmann condensation is a chemical reaction in which a 1,5-dicarbonyl compound reacts to form a cyclic β -keto ester. The reaction is named after the German chemist Hermann Dieckmann. The reaction involves the deprotonation of the α -carbon of one carbonyl group, followed by nucleophilic attack on the γ -carbon of the other carbonyl group, and subsequent cyclization and loss of water to form the cyclic β -keto ester product.

Diethyl ether - Wikipedia

Diethyl ether (C₄H₁₀O) is a colorless, flammable liquid with a characteristic sweet odor. It is commonly used as a solvent in organic chemistry and as an anesthetic. The molecule consists of two ethyl groups bonded to a central oxygen atom. It is a symmetrical ether.

Diethyl ether - Wikipedia

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