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Acid-base property of the resulting solution from a neutralization reaction depends on the remaining salt products. A salt containing reactive cations undergo hydrolysis by which they react with water molecules, causing deprotonation of the conjugate acids. For example, the acid salt ammonium chloride is the main species formed upon the half neutralization of ammonia in hydrochloric acid solution:

Acid salt - Wikipedia

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Fluorine is a chemical element with symbol F and atomic number 9. It is the lightest halogen and exists as a highly toxic pale yellow diatomic gas at standard conditions. As the most electronegative element, it is extremely reactive, as it reacts with almost all other elements, except for helium and neon.. Among the elements, fluorine ranks 24th in universal abundance and 13th in terrestrial ...

Fluorine - Wikipedia

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Lull, William P., with the assistance of Paul N. Banks. Conservation Environment Guidelines for Libraries and Archives. Ottawa, ON: Canadian Council of Archives, 1995. National Information Standards Organization. Environmental Guidelines for the Storage of Paper Records. Technical Report NISO-TR01-1995.

References â€“ Heritage Conservation Info

Formic acid is the simplest carboxylic acid. Formate is an intermediate in normal metabolism. It takes part in the metabolism of one-carbon compounds and its carbon may appear in methyl groups undergoing

transmethylation. It is eventually oxidized to carbon dioxide. Formate is typically produced as a byproduct in the production of acetate. It is responsible for both metabolic acidosis and ...

Formic acid | HCOOH - PubChem

The President of the European Academy of Sciences, Alain Tressaud and its Presidium invite you to the reception, organized on April 13th, 2018, at 11.30 a.m, to be held at the Fondation Universitaire in Brussels, for the occasion of the taking office of the new President Rodrigo Martins and the new Heads of Divisions and Officers.

Eurasc - News

Nitric acid (HNO₃). A colorless liquid that is used in the manufacture of inorganic and organic nitrates and nitro compounds for fertilizers, dye intermediates, explosives, and many different organic chemicals.

Nitric acid | HNO₃ - PubChem

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